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Synthesis and Biological Activity of Phospholipase C-Resistant Analogues of Phosphatidylinositol 4,5-bisphosphate

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The membrane phospholipid phosphatidylinositol 4,5-bisphosphate $(PtdIns(4,5)P_2)$ is an important regulator of cytoskeletal organization during a plethora of cellular functions, such as vesicle trafficking, endocytosis, phagocytosis, focal adhesion formation, and cell migration.¹ PtdIns(4,5)P₂ binds to and affects the function of many actin-binding and actin-remodeling proteins²⁻⁴ and is a cofactor in enzyme activation.⁵ In addition, PtdIns(4,5)P₂ regulates the activity of many ion channels and transporters.^{6,7} PtdIns(4,5)P₂ is also the source of three second messengers: $Ins(1,4,5)P_3$, diacylglycerol (DAG),^{8,9} and PtdIns(3,4,5)P₃.¹⁰ In many cases, it is the decrease in PtdIns(4,5)P₂, resulting from hydrolysis by phospholipase C (PLC) (Scheme 1), and not the increase in Ins- $(1,4,5)P_3$ and DAG that constitutes the physiologically relevant signal.^{11,12} Hydrolysis of PtdIns(4,5)P₂ causes TRP channels to lose some activity.^{13–19} Moreover, addition of PtdIns(4,5)P₂ restores sensitivity of TRPM4 and TRPM5 to activation by Ca2+ and restores the sensitivity of TRPM8 and TRPV1 to thermal and chemical stimuli.15,16,18,19

The availability of a metabolically stabilized analogue of PtdIns-(4,5)P₂, that is, one that lacks the scissile P–O bond and thus could not be hydrolyzed by PLC activity, would have many applications in understanding the role of PtdIns(4,5)P₂ in cell physiology. α -Fluoroalkylphosphonates have emerged as important nonhydrolyzable mimics for phosphoesters in the synthesis of biologically active "unnatural products".^{20–23} Herein we describe the first asymmetric total synthesis of isosteric and isoelectronic phosphonate analogues **1**–**5** of PtdIns(4,5)P₂ that cannot be hydrolyzed by PLC. The synthesis employs a Pd(0) coupling not previously exploited in phospholipid or phosphoinositide synthesis. Furthermore, we demonstrate that both saturated and unsaturated α -fluorophosphonate analogues can substitute for exogenous PtdIns(4,5)P₂ in restoring the sensitivity of the TRPM4 channel to Ca²⁺.

The synthetic sequence to the stabilized analogues **1**–**5** of PtdIns-(4,5)P₂ is illustrated in Scheme 2. A variety of attempts to connect the intermediate **9**²⁴ with a fluoromethylenephosphonic acid synthon²¹ failed. Eventually, we turned to the Pd(0)-catalyzed coupling of an *H*-phosphite with a vinyl bromide in order to form the desired C–P linkage. Thus, coupling the protected inositol **9** with dibenzyl *N*,*N*-diisopropylphosphoramidite gave the phosphoramidite intermediate **10**, which was converted to *H*-phosphonate **11** in 76% isolated yield in two steps.²⁵ The 1-bromo-1-fluoroolefin **7** (~1:1 *E/Z*) was separately prepared via a Et₂Zn-promoted olefination reaction²⁶ of CBr₃F/PPh₃ with glyceraldehyde **6** in excellent yield.

Few examples exist of Pd(0)-catalyzed formation of P–CF bonds, and in our hands, only traces of coupled compound **12** with a majority of the P–O cleaved compound **9** were obtained under standard conditions using Et₃N or K_2CO_3 as base. It appeared that

Scheme 1. Phospholipase C Catalyzes Hydrolysis of Ptdlns(4,5)P₂ to Two Second Messengers, $Ins(1,4,5)P_3$ and Diacylglycerol







^{*a*} Conditions: (a) CFBr₃, PPh₃, Et₂Zn, THF, 76%; (b) $(BnO)_2P(NPr_2-i)_2$, *N*,*N*-diisopropylethylammonium •1*H*-tetrazole, CH₂Cl₂, rt; (c) H₂O, 1*H*-tetrazole, rt, 1 h, CH₂Cl₂, 76% for two steps; (d) Pd(OAc)₂, dppf, propylene oxide, THF, 70 °C, 62%; (e) 60% aqueous TFA, THF, 0 °C, 1 h, 86%; (f) EDCI, DMAP, fatty acid, CH₂Cl₂, rt; (g) H₂, Pd/C, MeOH, 6 h, EtSH; (h) TMBr/TMSI (5:1), rt, 1.5 h; MeOH, 1 h.

the rate of decomposition was faster than the rate of coupling for the more hindered *H*-phosphonate **11.** To overcome this problem, we selected propylene oxide as a weak Lewis base and an effective scavenger of HBr.²⁷ Using this modification, treatment of the *H*-phosphonate **11** with Pd(OAc)₂/dppf/propylene oxide in THF at 70 °C led to the formation of α -fluorovinylphosphonate **12** in 62%

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Figure 1. PtdIns(4,5)P2 and analogues 2 and 4 restore TRPM4 currents following desensitization. (A) An excised inside-out patch from Chok1 cell expressing mouse TRPM4 (mTRPM4) shows activation and fast rundown of an inward current in the presence of 100 μ M Ca²⁺ and recovery by dioctanoyl-PtdIns(4,5)P₂ and analogues 2 and 4 ($V_m = 80 \text{ mV}$). (B) Initial magnitudes of the mTRPM4 currents, currents after rundown, and currents after recovery in response to 10 μ M each of PtdIns(4,5)P₂, 2, and 4 (averages, n = 8).

yield. Acetal 12 was selectively deprotected by treatment with 60% aqueous trifluoroacetic acid in tetrahydrofuran at 0 °C to give diol 13. Next, acylation of 13 with either octanoic acid, palmitic acid, or oleic acid provided the fully protected phosphonates 14a, 14b, and 14c in 80, 73, and 82% yields, respectively. Hydrogenolysis of 14a and 14b removed the benzyl groups, and then reaction with ethanethiol removed the MOM groups to give the α -fluoromethylenephosphonate analogues 1 and 2.²⁸ The α -fluorovinylphosphonates $3-5^{28}$ were obtained by deprotection of benzyl and MOM groups simultaneously with TMSBr/TMSI (5:1).

Recently, the hydrolysis of the water-soluble dioctanoyl PtdIns- $(4,5)P_2$ was found to be important in the desensitization of TRPM4 channel (activated by cytoplasmic Ca²⁺). Exogenous PtdIns(4,5)- P_2 could restore the sensitivity of TRPM4 channels to Ca^{2+} , demonstrating that PtdIns(4,5)P₂ was a general regulator for the gating of TRPM4 ion channels.15 The ability of the two dioctanoyl-PtdIns(4,5)P₂ analogues 2 and 4 to restore TRPM4 currents following rundown is shown in Figure 1. Both analogues restored TRPM4 sensitivity following desensitization, but the α -fluorovinylphosphonate 4 was more potent. Indeed, the unsaturated phosphonate 4 was even more effective than the hydrolyzable dioctanoyl-PtdIns(4,5)P₂ at restoring TRPM4 sensitivity. This provides further evidence that the regulation of TRPM4 by dioctanoyl-PtdIns(4,5)-P2 and the ability of dioctanoyl-PtdIns(4,5)P2 to restore TRPM4 currents following rundown is not due to effects of products of PLC hydrolysis.15

To determine sensitivity of TRPM4 currents to 2 and 4, we measured the effects of varying concentrations of both compounds on the recovery of TPRM4 currents in excised inside-out patches evoked in response to 100 μ M Ca²⁺ (Figure 2). Maximal recovery of TRPM4 currents was observed upon reaching 10 μ M for both 2 and 4, and half-activation was observed at $\sim 2 \ \mu M$ for both compounds, which is similar to the concentration of PtdIns(4,5)P₂ that promoted half-activation of TRPM4 (6 µM).¹⁵ The difference between the effectiveness of 2 and 4 in restoring TRPM4 currents (Figure 1) appears to result from differential abilities to promote activation of the TRPM4 channel. Taken together, these data suggest that the α -fluorovinylphosphonate 4 is a biologically active, longlived mimic of PtdIns(4,5)P₂.

In conclusion, we developed an efficient synthesis of two nonhydrolyzable PtdIns(4,5)P2 analogues, and we showed that α -fluorovinylphosphonate 4 optimally restored the sensitivity of



Figure 2. Dose-response for recovery of TRPM4 currents by 2 and 4. After TRPM4 desensitization, recovery was assessed. Data were normalized to the response to 10 μ M of each analogue in the same patch. (A) Averaged data (n = 5) for recovery of TRPM4 currents by **2** (EC₅₀ = 2.7 ± 0.6 μ M and $n_{\rm H} = 2.5 \pm 1.2$). (B) Averaged data (n = 6) for 4 (EC₅₀ = 1.8 ± 0.1 μ M and $n_{\rm H} = 3.2 \pm 0.5$).

TRPM4 currents. These results suggest that metabolically stabilized analogues of PtdIns(4,5)P₂ will have a wide variety of applications in separating the role of the phosphoinositide per se from activities that result when Ins(1,4,5)P₃, DAG, Ca²⁺, or other downstream signals are generated from the hydrolysis of $PtdIns(4,5)P_2$ by PLC.

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Supporting Information Available: Experimental details for synthesis and characterization of new compounds, and protocols for TRPM4 channel activity measurement. This material is available free of charge via the Internet at http://pubs.acs.org.

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